

Atomic Structure of Graphene on SiO₂

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ABSTRACT

We employ scanning probe microscopy to reveal atomic structures and nanoscale morphology of graphene-based electronic devices (i.e., a graphene sheet supported by an insulating silicon dioxide substrate) for the first time. Atomic resolution scanning tunneling microscopy images reveal the presence of a strong spatially dependent perturbation, which breaks the hexagonal lattice symmetry of the graphitic lattice. Structural corrugations of the graphene sheet partially conform to the underlying silicon oxide substrate. These effects are obscured or modified on graphene devices processed with normal lithographic methods, as they are covered with a layer of photoresist residue. We enable our experiments by a novel cleaning process to produce atomically clean graphene sheets.

Graphene,^{1,2} a single layer of graphite, is an unique material with exotic electronic properties.^{1–8} A hexagonal two-dimensional network of carbon atoms composes graphene; it is exactly one atom in thickness, and every carbon atom is a surface atom. Therefore, substrate-induced structural distortion,⁹ adsorbates,⁷ local charge disorder,¹⁰ atomic structure at the edges,^{4,11} and even atomic scale defects¹² could be very important for transport properties of graphene. Specifically, lowered carrier mobility⁹ and suppression of weak localization⁹ in graphene-based devices have been attributed to corrugation of the graphene. Consequently, understanding the atomic and nanoscale structures of graphene in the configuration in which it is measured is crucial to explaining the observed transport properties.

Experimentally, controlling the environment of graphene in a device configuration is difficult. Graphene on the common gate dielectric, SiO₂, is subject to the effects of trapped oxide charges,¹³ which are highly dependent on sample preparation. In addition, graphene devices are typically fabricated using electron beam lithography, exposing the graphene to photoresist that can leave behind contaminants which, like any chemical adsorbate, may modify electronic transport properties,¹⁰ may play a large role in reported graphene response to gas exposure,⁷ etc. For instance, a free-standing graphene sheet has been reported to have intrinsic three-dimensional (3D) structure or ripples due to the instability of two-dimensional (2D) crystals.^{14,15} However, the structures characterized had been exposed to photoresist, leaving the possibility that effects of chemical

residues may have influenced the observed structure. Carefully controlling the experimental variables such as the influence of the substrate and the presence of impurities is necessary to interpret observed transport properties correctly.

In this Letter, we report atomic structure and nanoscale morphology of monolayer graphene sheets and nanotubes in the most commonly used device configurations (i.e., on an insulating SiO₂ substrate with conducting back gate and fabricated electrical contacts). We find that acrylic lithography resists, commonly used in the device fabrication procedure, introduce unknown and uncontrollable perturbations, which must also apply to the majority of previously reported graphene devices. The removal of the residue is necessary for uncovering intrinsic structural properties of the graphene sheet. Upon removing the resist residue, we are able to acquire atomic-resolution images of the graphene lattice, which shows both triangular and hexagonal lattice patterns in close proximity, indicating significant scattering of the electron waves. The atomic-resolution images also prove that our graphene devices are clean to atomic scale, enabling controlled analysis of the structural properties. Finally, we measure the thickness of a graphene film in ultrahigh vacuum (UHV) and in ambient and show that the large height measured in ambient is due to significant presence of atmospheric species under and/or on the graphene film.

We use scanning tunneling microscopy (STM) to achieve atomic-scale resolution, while we compare nanoscale morphologies of graphene and silicon dioxide substrate by noncontact atomic force microscopy (AFM). Unless otherwise noted, our microscopy studies were performed in UHV. In graphene devices, only the electronic contacts and

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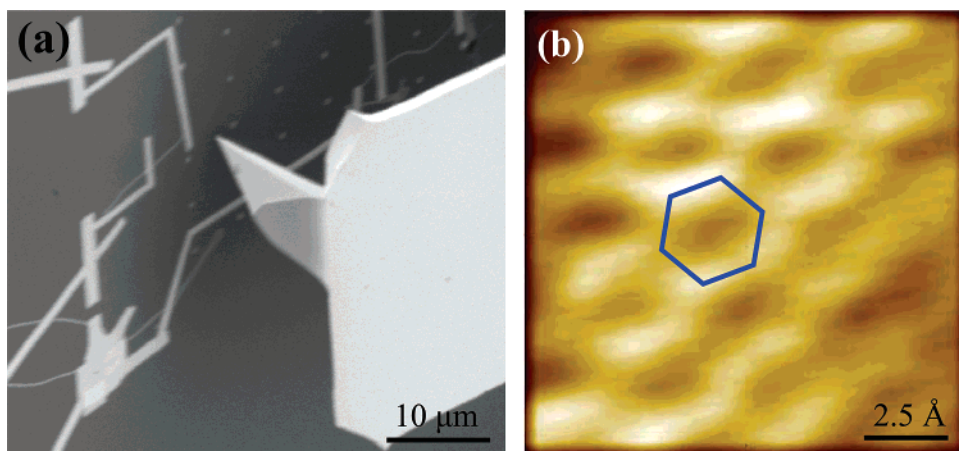


Figure 1. (a) Scanning electron micrograph of a carbon nanotube device, showing our experimental setup. The triangular shape to the right of the image is the tip of the scanning probe. The nanotubes were grown using chemical vapor deposition following ref 32, and the electrodes were patterned using a standard two-step electron-beam lithography process.³³ The device substrate is 500 nm thick thermal SiO₂ grown on a heavily doped silicon wafer. Wide, near vertical lines on the left are electrical contacts. Thin white lines are the nanotubes lying on the surface of SiO₂. (b) An STM image of a nanotube in the device configuration, showing atomic structure. $V_{\text{source}} = V_{\text{drain}} = 1.4$ V, $V_{\text{gate}} = 0$ V, and $I_{\text{tunnel}} = 18$ pA.

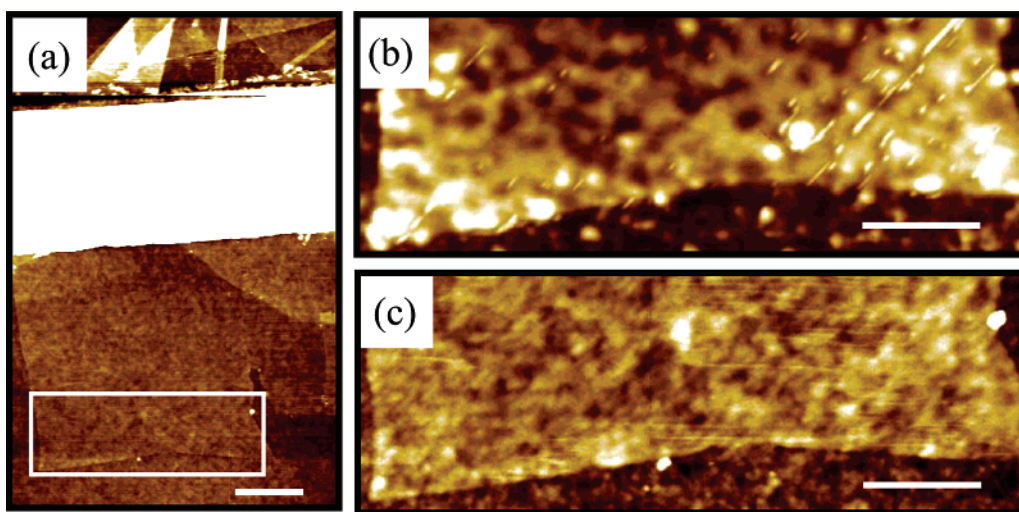


Figure 2. (a) AFM topography of graphene deposited on SiO₂. Thin graphite flakes are generated using the mechanical exfoliation technique¹ on thermally grown SiO₂ with the thickness of 300 nm. Monolayer graphite flakes (graphene) are located using optical and atomic force microscopy.⁸ The e-beam lithography defined electrode,³³ approximately 80 nm in height and 1.5 μm in width, is the white area nearly horizontal to the image. The black square indicates the region shown in parts b and c of Figure 1. The scale bar is 500 nm. (b) Graphene sheet prior to the cleaning procedure described in text. The scale bar is 300 nm. (c) Graphene sheet after the cleaning procedure. The standard deviation of the height variation in a square of side 600 nm is approximately 3 Å after the treatment compared to 8 Å before the treatment. The scale bar is 300 nm. Images a–c were acquired using intermittent-contact mode AFM in air.

graphene are conducting while the gate dielectric, which is insulating, composes the vast majority of the device substrate. Since STM requires conductive substrates, the STM tip must be positioned exactly above only the conductive areas, which extend laterally only several nanometers to micrometers for graphene devices. We use a commercial UHV system,¹⁶ which features a field emission scanning electron microscope (SEM) combined with AFM and STM for rapid, reproducible placement of scanned probe. Figure 1a is an SEM image showing the scanned-probe tip approaching a representative carbon nanotube device to demonstrate our tip placement capability. In Figure 1a, the nanotubes appear as thin curved white lines and the source/drain electrodes as the wider near-vertical lines, and a conductive AFM cantilever¹⁷ is visible

on the right. Coarse positioning of the cantilever within several micrometers of the nanotube is performed using SEM. We then utilize noncontact frequency-shift AFM¹⁸ to locate the nanotube and to place the cantilever within several nanometers of the nanotube. Finally, the cantilever is employed as the STM tip; the tunneling current travels from the cantilever into the nanotube and along the nanotube into the electrodes. STM imaging is limited to the nanotube. As shown in Figure 1b, this integrated technique is successful in resolving the atomic structure of nanotubes in the device configuration.

Figure 2a is an AFM image of the graphene-based device, which we discuss in this paper. The wide white line, approximately 1 μm in width, is one electrode. The contacted

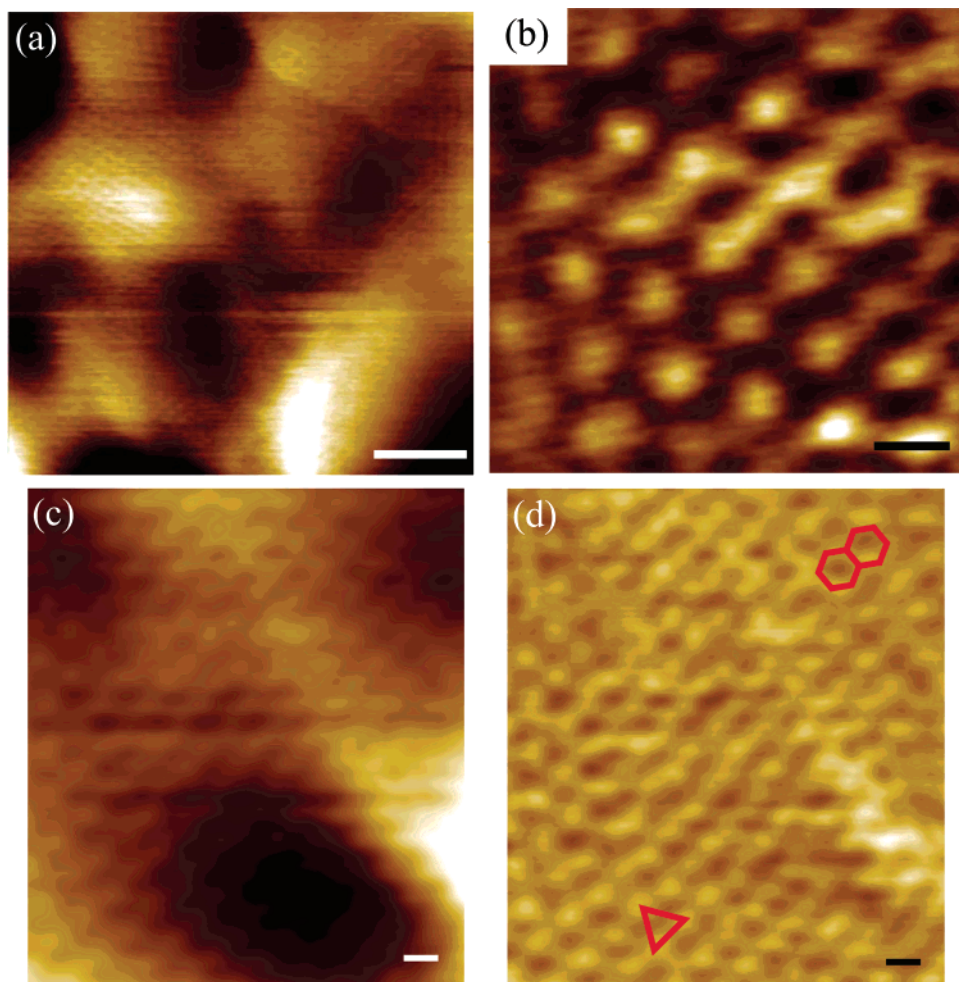


Figure 3. (a) A typical large-area STM image of the graphene sheet shown in Figure 2a. Peak-to-peak height variation of the image is approximately 2.5 nm. $V_{\text{sample}} = 1.1$ V and $I_{\text{tunnel}} = 0.3$ nA. The scale bar is 2 nm. (b) Atomically resolved image of a graphene sheet. $V_{\text{sample}} = 1.0$ V and $I_{\text{tunnel}} = 24$ pA. The scale bar is 2.5 Å. (c) STM image of another area. The scale bar is 2.5 Å. $V_{\text{sample}} = 1.2$ V and $I_{\text{tunnel}} = 0.33$ nA. (d) A high-pass filtered image of the large area scan shown in (c). Both triangular and hexagonal patterns are observed. The orientations of the red triangle and hexagons are same. The scale bar is 2.5 Å.

graphitic material varies in the thickness but the large section appearing to the lower left is uniformly one monolayer thick, as will be shown later.

We find that a continuous film covers the surface of the graphene devices after the lift-off procedure, and it is not possible to obtain atomic resolution images via STM. A similar film was seen on the majority of nanotube segments in the nanotube devices, with only localized clean segments suitable for imaging. A control experiment using the same resist deposition¹⁹ and acetone resist liftoff procedures on highly oriented pyrolytic graphite yields the same film, confirming its origin as residue from the resist. This indicates that the resist residue covers all graphene devices fabricated using similar photoresist process. Standard solvents such as Nano Remover PG²⁰ and glacial acetic acid do not perturb the residue. Known resist cleaning processes are inadequate for completely removing the resist residue.

We are able to remove the photoresist residue in argon/hydrogen atmosphere at 400 °C.²¹ Figure 2c shows the AFM image of the same area shown in Figure 2b, after the heat treatment. The graphene sheet now appears with finer,

smoother corrugations. A representative large-area STM image of the cleaned graphene sheet is shown in Figure 3. The atomic-scale pattern is visible in Figure 3a and can be imaged clearly at higher resolution as shown in parts b and d of Figure 3. The meandering of atomic rows seen in Figure 3d is due to the curvature of the surface.²² The observed lattice spacing is consistent with the graphene lattice, and the appearance of both triangular and hexagonal lattice in the image indicates the presence of strong spatially dependent perturbations which interact with graphene electronic states.^{23,24} Such perturbations may be due to the observed film curvature and/or the charge traps on the SiO₂ surface. Significantly, STM images at any position on the device always reveal the graphitic lattice. Therefore, surface impurities have been removed completely from the graphene surface, and the corrugation seen in Figure 2c is representative of the clean graphene sheet on SiO₂.

The material thickness is one of the key structural factors in determining the properties of graphene-based devices.³ Figure 4a shows an AFM image of the boundary between the same graphene sheet and SiO₂ substrate. A histogram

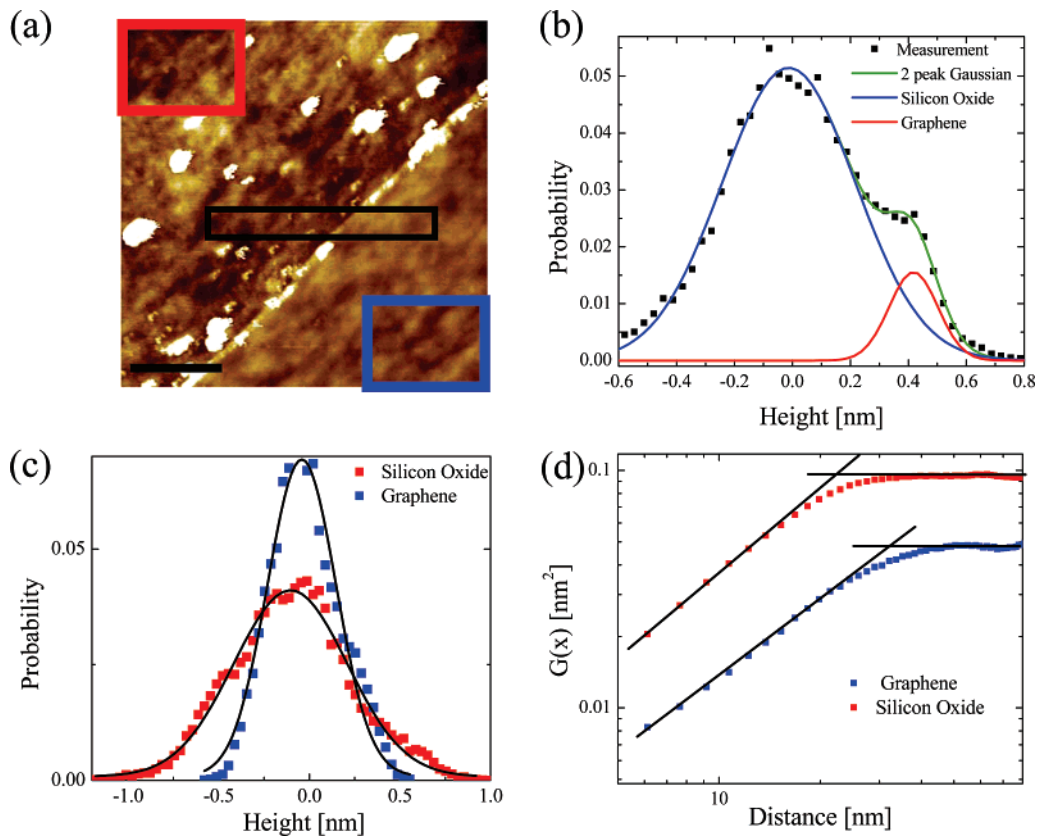


Figure 4. (a) Noncontact mode AFM image, acquired in UHV, of a boundary between the graphene sheet and SiO₂ substrate. The graphene sheet occupies the lower right area of the image. The scale bar is 200 nm. The black rectangle indicates the area for the histogram shown in Figure 4b, and red and blue rectangles indicate the area where the histograms shown in Figure 4c has been acquired. (b) Height histogram acquired across the graphene–substrate boundary (black rectangle in Figure 4a). The data are fit by two Gaussian distributions (solid red and blue lines; green line is sum), with means separated by 4.2 Å. (c) Height histograms acquired on graphene and SiO₂ (red and blue squares, respectively, in Figure 4a). The histograms are well-described by Gaussian distributions (black lines) with standard deviations of 1.9 and 3.1 Å for graphene and SiO₂, respectively. (d) The height–height correlation function (see text) of the graphene sheet and SiO₂ surface. The lines are fits to the large and small length behaviors (power-law and constant, respectively), and the point of intersection indicates the correlation length. This analysis is performed by selecting data from Figure 4a, showing both graphene and SiO₂ surfaces. Therefore, the tip morphology is the same for both curves and the tip-related artifact effect does not contribute to the analysis.

acquired across the boundary shown in Figure 4b shows that the film thickness is 4.2 Å, comparable to the layer-to-layer spacing in bulk graphite of 3.4 Å. Therefore, the imaged graphene device area is a monolayer. Similar analysis performed in air for the same area, before our experiments in UHV, shows the thickness to be 9 Å, consistent with a previous measurement of a monolayer material in air.^{1,8} The discrepancy between the air/vacuum measurements of 4.6 Å in thickness indicates a significant presence of ambient species (nitrogen, oxygen, argon, or water) between SiO₂ and the graphene sheet and/or on the graphene sheet.

We now turn our attention to the 3D morphology of the graphene sheet, important for the transport properties.⁹ Figure 4c shows histograms of the heights over graphene and SiO₂. The graphene sheet is approximately 60% smoother than the oxide surface; the standard deviations of the measured height variations are 1.9 and 3.1 Å for the graphene and oxide surface. The height–height correlation function, $g(x) = \langle (z(x_0 + x) - z(x_0))^2 \rangle$, is a useful measure for characterizing the surface morphology.^{25–27} Figure 4d shows the height–height correlation function²⁸ for the graphene and SiO₂ surface. Both correlation functions rapidly increase as $g \sim$

x^{2H} at short distances, as expected:²⁶ $2H = 1.11 \pm 0.013$ for graphene and $2H = 1.17 \pm 0.014$ for SiO₂. A value of the exponent $2H \sim 1$ indicates a domain structure with short-range correlations among neighboring domains²⁵ and is not surprising for SiO₂. A value of $2H = 2$ is expected²⁹ for a thermally excited flexible membrane under the influence of an interaction (e.g., van der Waals) with the substrate. Consequently, the observed $2H$ value demonstrates that the observed graphene morphology is not representative of the intrinsic structure. A rollover at the correlation length and saturation at mean square roughness at large distances follow the short-distance behavior. As seen in the figure inset, interpolating the intersection of the power law and saturated regimes yields values of the correlation length,²⁶ which are $\xi = 32 \pm 1$ nm for graphene and $\xi = 23 \pm 0.6$ nm for SiO₂. The similar exponents and slightly larger correlation length of the graphene sheet are consistent with the graphene morphology being determined by the underlying SiO₂ substrate. The larger correlation length and smaller roughness of the graphene surface would arise naturally due to an energy cost for graphene to closely follow sharp orientation changes on the substrate. Free-standing graphene has been

reported to have larger static nanoscale corrugations^{14,15} attributed to intrinsic structural instability of 2D materials. However, the free-standing graphene was treated with a resist process,^{14,15} and the resulting resist residue could certainly prevent the graphene sheet from reaching its equilibrium structural corrugation.

The observed corrugations in our study indicate a maximum local strain of approximately 1%. Using the Young's modulus of 1 TPa³⁰ and graphene thickness of 3.4 Å, the corresponding stored energy density due to the induced deformation is ~ 1 meV/Å². We estimate the graphene–SiO₂ interaction energy to be 6 meV/Å² based on the interlayer van der Waals interaction in graphite³¹ of 20 meV/Å² at the distance of 3.4 Å. The estimated interaction energy between the graphene sheet and SiO₂ substrate is thus sufficient to overcome the energy cost of the corrugations needed for graphene to follow the SiO₂ morphology.

Corrugations comparable to those observed here have been postulated to be responsible for the lack of low-field magnetoresistance observed in graphene on SiO₂ via suppression of weak localization due to the introduction of an effective random magnetic field.⁹ Indeed, “flatter” graphene films, prepared on SiC with the film coherence length of 90 nm, show weak localization.² The corrugations in graphene on SiO₂ were later attributed to intrinsic corrugations in the graphene itself.¹⁵ However, our findings indicate that the graphene corrugations that are relevant for interpreting many reports of device performance (e.g. for graphene on SiO₂) are due to partial conformation of the graphene to the SiO₂, not to the intrinsic corrugation of graphene.

We have resolved atomic structures of oxide-supported graphene-based electronic devices using a novel combined SEM/AFM/STM technique. We obtain real-space images of the single-layer graphene atomic lattice for the first time and characterize the thicknesses and nanoscale corrugation of a clean graphene sheet devoid of any impurities. Our observation shows that the graphene primarily follows the underlying morphology of SiO₂ and thus does not have intrinsic, independent corrugations on SiO₂. The graphene sheets *do* have finite intrinsic stiffness, which prevents the sheets from conforming completely to the substrate. In addition, we demonstrate that resist residues are ubiquitous on lithographically fabricated graphene devices, and their presence should be considered in interpreting transport and structural measurements of earlier studies. Our quantitative measure of the extrinsic corrugations of graphene on SiO₂ can be used as input to theoretical models of strain-induced disorder in graphene and its effect on transport properties. Furthermore, our observation that graphene can conform to substrate morphology suggests new experimental directions: the use of controlled substrate morphologies (e.g., a patterned SiO₂ substrate, or alternative dielectric materials) may be a useful approach to investigate how the corrugation-induced strain impacts the transport properties of graphene. Finally, our technique (the novel integrated microscopy allied with the resist cleaning process) can be applied to resolve atomic structures of nanoelectronic devices in general; the technique

finally enables studies of the impact of atomic scale defects and adsorbates on nanoscale transport properties.

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References

- (1) Novoselov, K. S.; et al. Two-dimensional atomic crystals. *Proc. Natl. Acad. Sci. U.S.A.* **2005**, *102*, 10341–10453.
- (2) Berger, C.; et al. Electronic confinement and coherence in patterned epitaxial graphene. *Science* **2006**, *312*, 1191–1196.
- (3) Castro, E. V.; et al. Biased bilayer graphene: semiconductor with a gap tunable by electric field effect. preprint archive, xxx.lanl.gov. 2007.
- (4) Nakada, K.; Fujita, M.; Dresselhaus, G.; Dresselhaus, M. S. Edge state in graphene ribbons: nanometer size effect and edge shape dependence. *Phys. Rev. B* **1996**, *54*, 17954–61.
- (5) Novoselov, K. S.; et al. Electric field effect in atomically thin carbon films. *Science* **2004**, *306*, 666–669.
- (6) Novoselov, K. S. G. A. K.; Morozov, S. V.; Jiang, D.; Katsnelson, M. I.; Grigorieva, I. M.; Dubonos, S. V.; Firsov, A. A. Two-dimensional gas of massless Dirac fermions in graphene. *Nature* **2005**, *438*, 197–200.
- (7) Schedin, F.; Novoselov, K. S.; Morozov, S. V.; Jiang, D.; Hill, E. H.; Blake, P.; Geim, A. K. Detection of individual gas molecules by graphene sensors. preprint archive, xxx.lanl.gov, cond-mat/0610809, 2007.
- (8) Zhang, Y.; Tan, Y.-W.; Stormer, H. L.; Kim, P. Experimental observation of the quantum Hall effect and Berry's phase in graphene. *Nature* **2005**, *438*, 201–204.
- (9) Morozov, S. V.; et al. Strong suppression of weak localization in graphene. *Phys. Rev. Lett.* **2006**, *97*, 016801.
- (10) Huang, E. H.; Adam, S.; Das Sarma, S.; Geim, A. K. Transport in chemically doped graphene in the presence of adsorbed molecules. xxx.lanl.gov preprint archive cond-mat/0610834 2006.
- (11) Niimi, Y.; et al. Scanning tunneling microscopy and spectroscopy of the electronic local density of states of graphite surfaces near monoatomic step edges. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2006**, *73*, 85421–851.
- (12) Wehling, T. O. et al. Local electronic signatures of impurity states in graphene. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2007**, *75*, 125425.
- (13) Jayaraman, R.; Sodini, C. G. A 1/f noise technique to extract the oxide trap density near the conduction band edge of silicon. *IEEE Trans. Electron Devices* **1989**, *36*, 1773.
- (14) Meyer, J. C.; et al. The structure of suspended graphene sheets. *Nature* **2007**, *60*, 446.
- (15) Meyer, J. C.; et al. On the roughness of single- and bi-layer graphene membranes. Preprint archive, xxx.lanl.gov, cond-mat/0703033, 2007.
- (16) The microscope was manufactured by the Japanese Electro-Optics Laboratory. We have significantly modified the sample bias voltage circuitry and the sample holder to allow three independent biasing electrodes for our measurement.
- (17) Noncontact mode AFM was performed using NSC 15 Rectangular cantilever (325 kHz, 40 N/m) from MicroMasch.
- (18) *Noncontact Atomic Force Microscopy*; Morita, S., Wiesendanger, R., Meyer, E., Eds.; Springer-Verlag: Berlin, 2002.
- (19) For this control experiment, we tested both poly(methyl methacrylate) (PMMA) and copolymer PMMA/8.5% methacrylic acid (MMA-MAA) resists in contact with the substrate. Both resists show similar resist residues upon lift-off.
- (20) *N-Methyl-2-pyrrolidone (NMP)*-based remover sold by Microchem.
- (21) The flow rates were 1700 mL/min for argon and 1900 mL/min for hydrogen. Our anneal time was 1 h. Ar/H₂ flow must be maintained throughout the heating and annealing. Otherwise, the residue film is not removed.

- (22) Venema, L. C.; Menueir, V.; Lambin, P.; Dekker, C. Atomic structure of carbon nanotubes from scanning tunneling microscopy. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2000**, *61*, 2991.
- (23) Tomanek, D.; et al. Theory and observation of highly asymmetric atomic structure in scanning-tunneling-microscopy images of graphite. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1987**, *35*, 7790.
- (24) Kane, C. L.; Mele, E. J. Broken symmetries in scanning tunneling images of carbon nanotubes. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1999**, *59*, R12759.
- (25) Goldberg, J. L.; Wang, X.-S.; Bartelt, N. C.; Williams, E. D. Surface height correlation functions of vicinal Si(111) surfaces using scanning tunneling microscopy. *Surf. Sci. Lett.* **1991**, *249*, L285–92.
- (26) Krim, J.; Heyvaert, I.; Haesendonck, C. V.; Bruynseraede, Y. Scanning Tunneling Microscopy Observation of Self-Affine Fractal Roughness in Ion-Bombarded Film Surfaces. *Phys. Rev. Lett.* **1993**, *70*, 57.
- (27) Palasantzas, G.; Tsamoras, D.; De Hosson, J. T. M. Roughening aspects of room temperature vapor deposited oligomer thin films onto Si substrate. *Surf. Sci.* **2002**, *507–10*, 357–361.
- (28) For each correlation function, we average seven different horizontal line profiles acquired on each respective surface. Constant linear offsets at large distances from 50 to 150 nm have been subtracted to account for imperfect plane tilt correction performed on the image.
- (29) Aranda-Espinoza, H.; Lavallee, D. Structure factor of flexible membranes. *Europhys. Lett.* **1998**, *43*, 355–359.
- (30) Krishnan, A.; Dujardin, E.; Ebbesen, T. W.; Yianilos, P. N.; Treacy, M. M. J. Young's modulus of single-walled nanotubes. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1998**, *58*, 14013–14019.
- (31) Chopra, N. G.; et al. Fully collapsed carbon nanotubes. *Nature* **1995**, *377*, 135–138.
- (32) Kim, W. K.; et al. Synthesis of ultralong and high percentage of semiconducting single-walled carbon nanotubes. *Nano Lett.* **2002**, *2*, 703.
- (33) In the first electron-beam lithography (EBL) step, alignment markers are defined using EBL near the monolayer materials. Electronic contacts are then defined in the second EBL step using the alignment markers. The contacts are gold with chromium sticking layer. Typically, PMMA/MMA-MAA bilayer resist is used for EBL.

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